



Porphyrins and bile pigments: metabolism and disorders

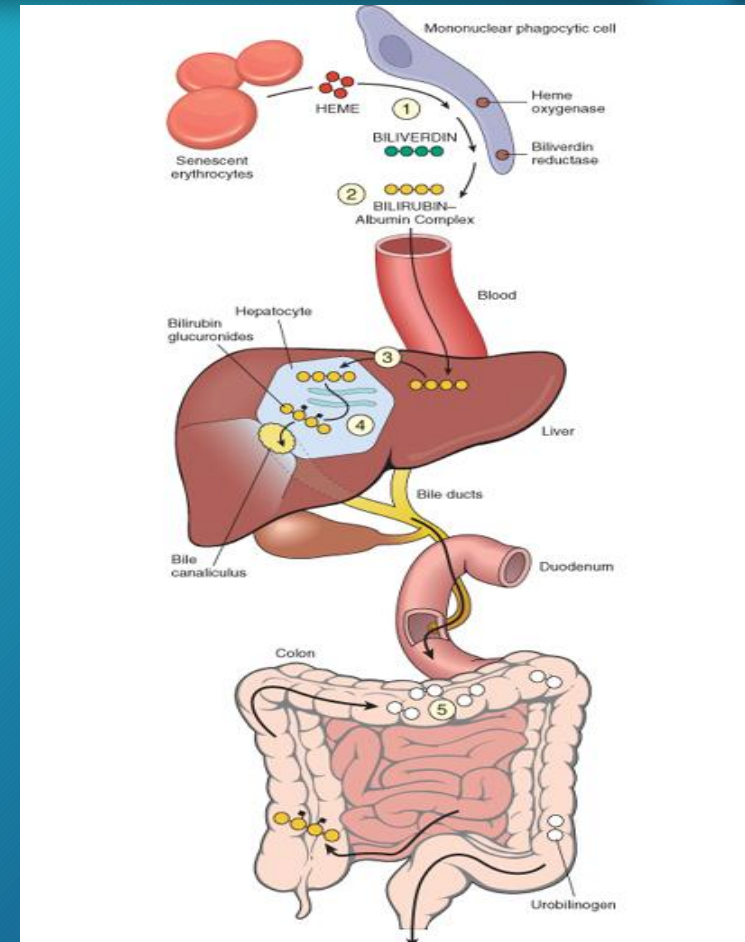
Dr. Jaya Chaturvedi

Porphyrins

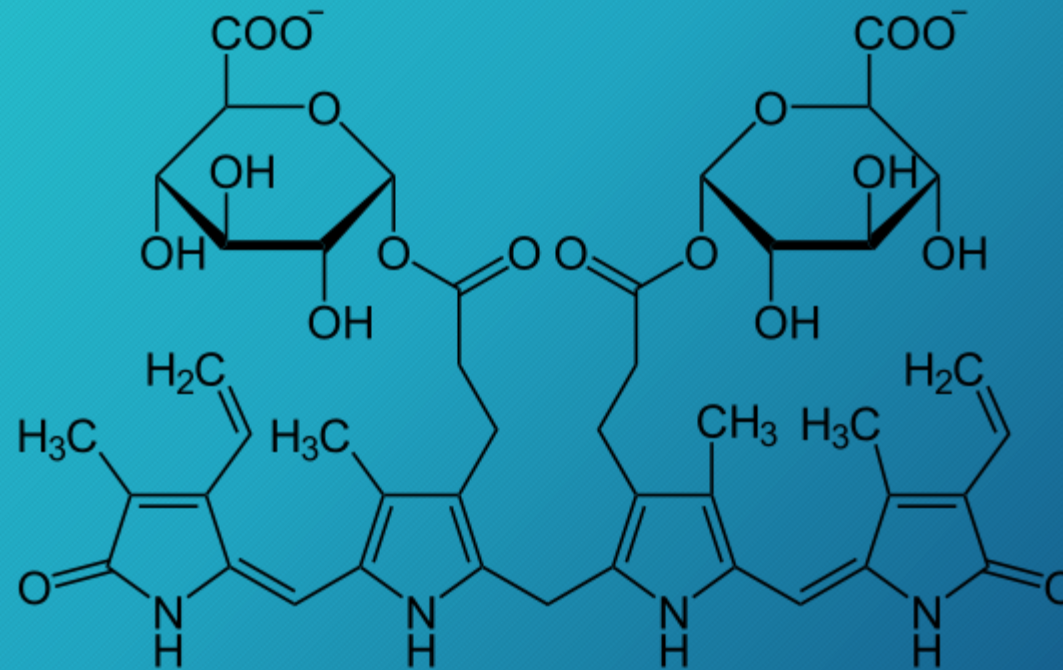
- Porphyrins are cyclic compounds formed by the linkage of four pyrrole rings through methyne (—C—C—) bridges. In the naturally occurring porphyrins, various side chains replace the eight numbered hydrogen atoms of the pyrroles.
- Porphyrins have had different structures depend on side chains that are attached to each of the four pyrrole rings. For example; Uroporphyrin, coporphyrin and protoporphyrin IX (heme).
- The most prevalent metalloporphyrin in humans is heme, which consists of one ferrous (Fe^{2+}) iron ion coordinated at the center of the tetrapyrrole ring of protoporphyrin IX.

What is bilirubin?

- Bilirubin is a yellowish pigment found in bile, a fluid made by the liver.
- The breakdown product of Hgb from injured RBCs and other heme containing proteins.
- Produced by reticuloendothelial system
- Released to plasma bound to albumin
- Hepatocytes conjugate it and excrete through bile channels into small intestine.

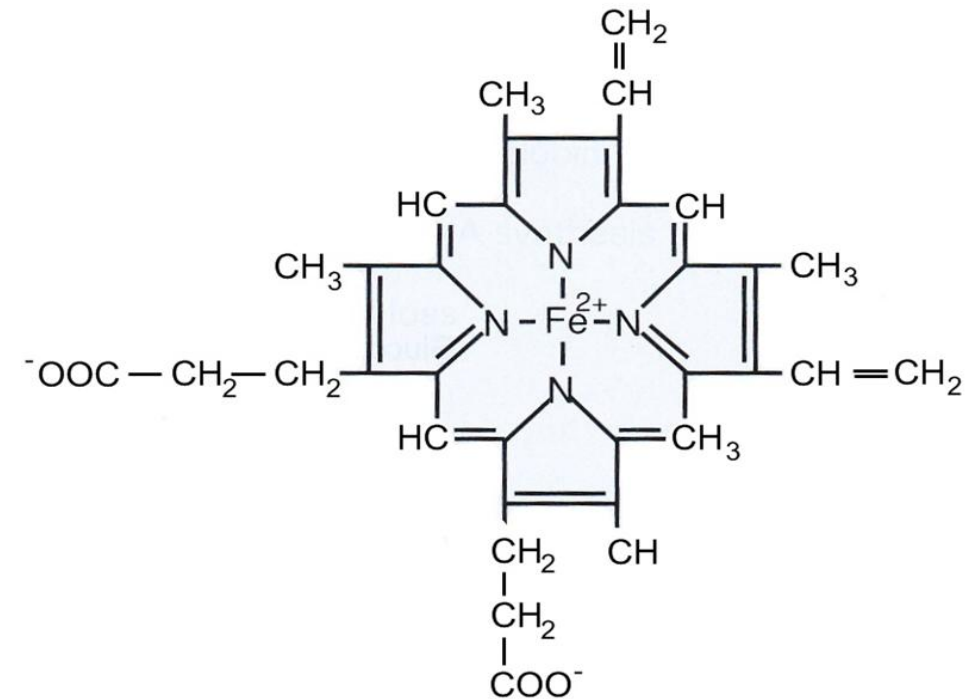


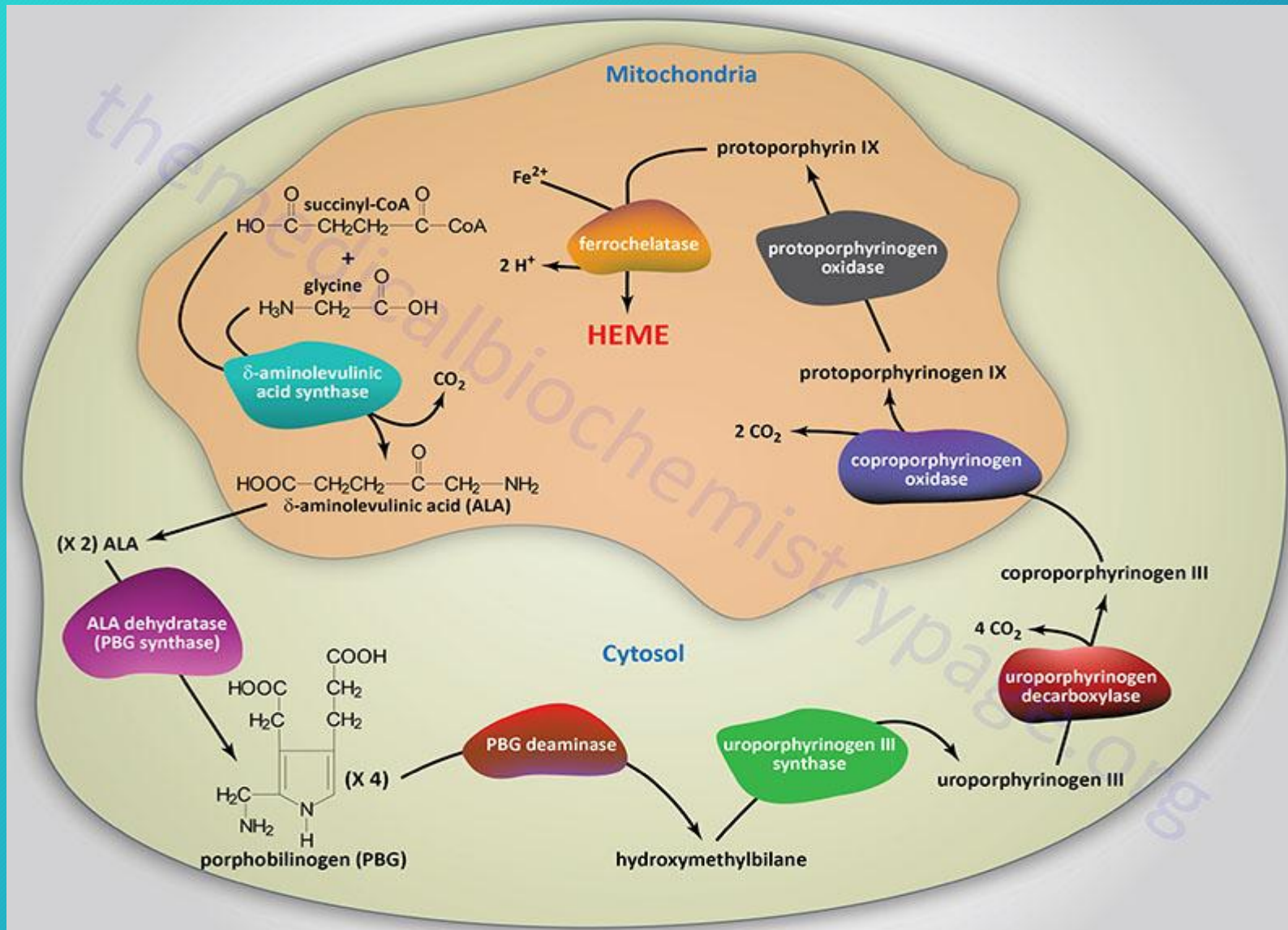
Bilirubin di-glucoronid



Structure of heme:

- **Heme structure:**
 - a porphyrin ring coordinated with an atom of iron
 - side chains: methyl, vinyl, propionyl
- **Heme is complexed with proteins to form:**
- Hemoglobin, myoglobin and cytochromes



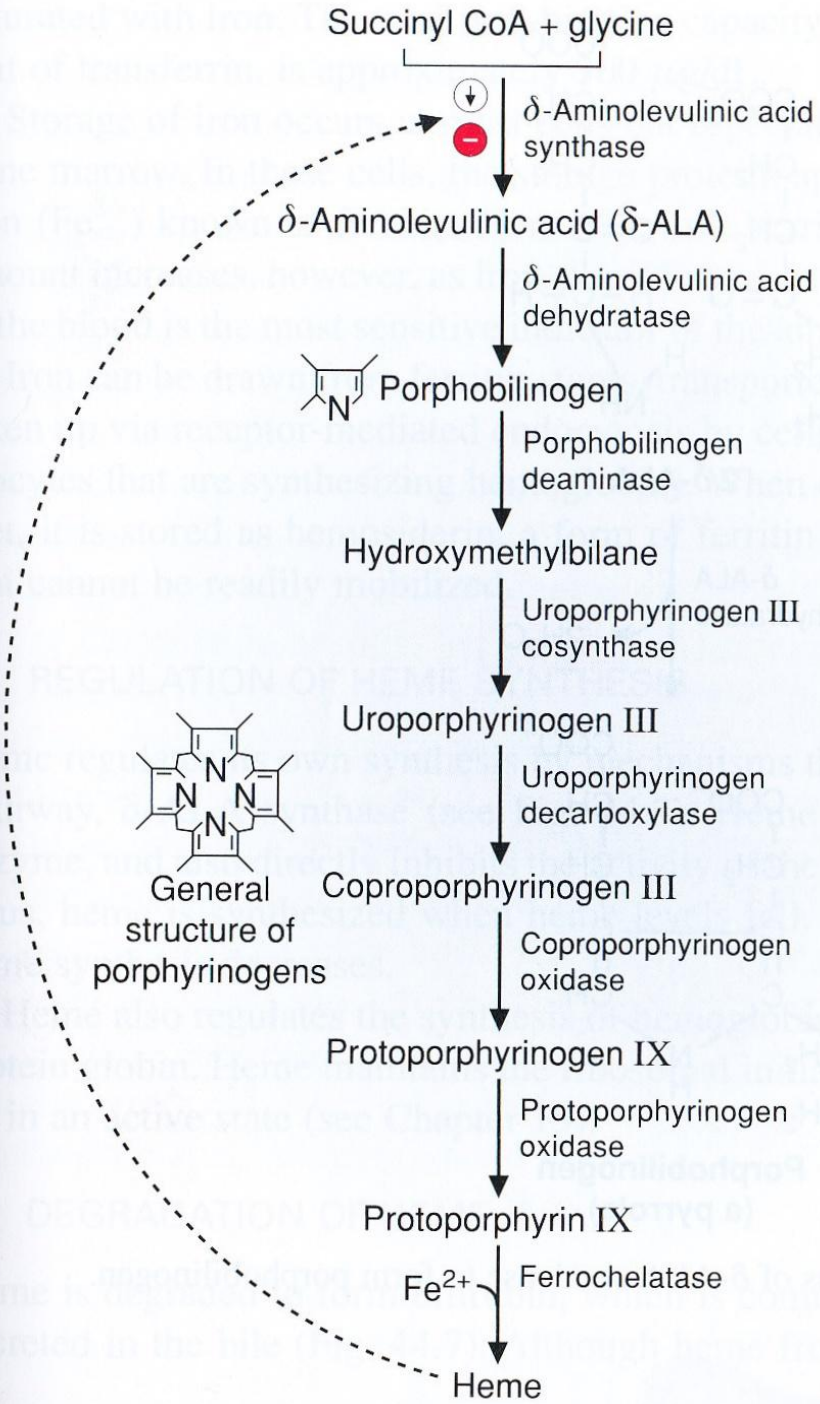


Pathway of Heme Biosynthesis. Heme biosynthesis begins in the mitochondria from glycine and succinyl-CoA, continues in the cytosol, and ultimately is completed within the mitochondria. The heme that it produced by this biosynthetic pathway is identified as heme *b*. PBG: porphobilinogen; ALA: δ-aminolevulinic acid

Mitochondria

Cytosol

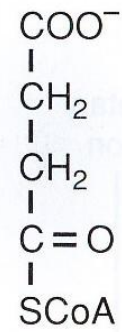
Mitochondria



Porphyrias

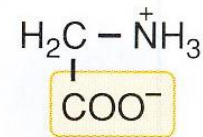
- δ -ALA dehydratase porphyria
- Acute intermittent porphyria
- Congenital erythropoietic porphyria
- Porphyria cutanea tarda
- Hereditary coproporphyria
- Variegate porphyria
- Erythropoietic protoporphyria

Synthesis of δ -aminolevulinic acid:

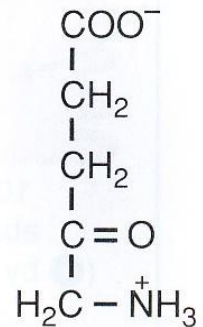
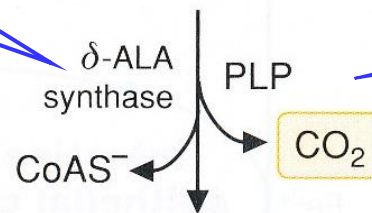


Succinyl CoA

+



Glycine

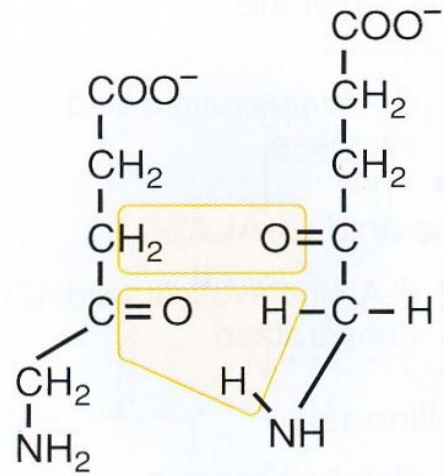


δ -Aminolevulinic acid
(δ -ALA)

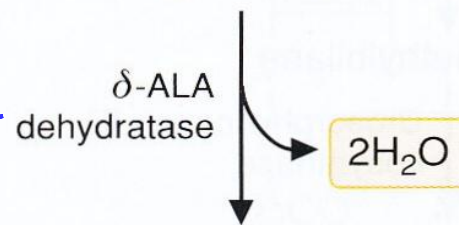
induced by: drugs (barbiturates),
oral contraceptive pills

Pyridoxal phosphate
(vit. B₆)

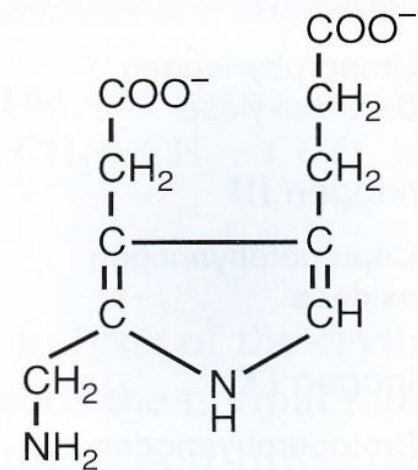
Formation of porphobilinogen:



2 δ -ALA

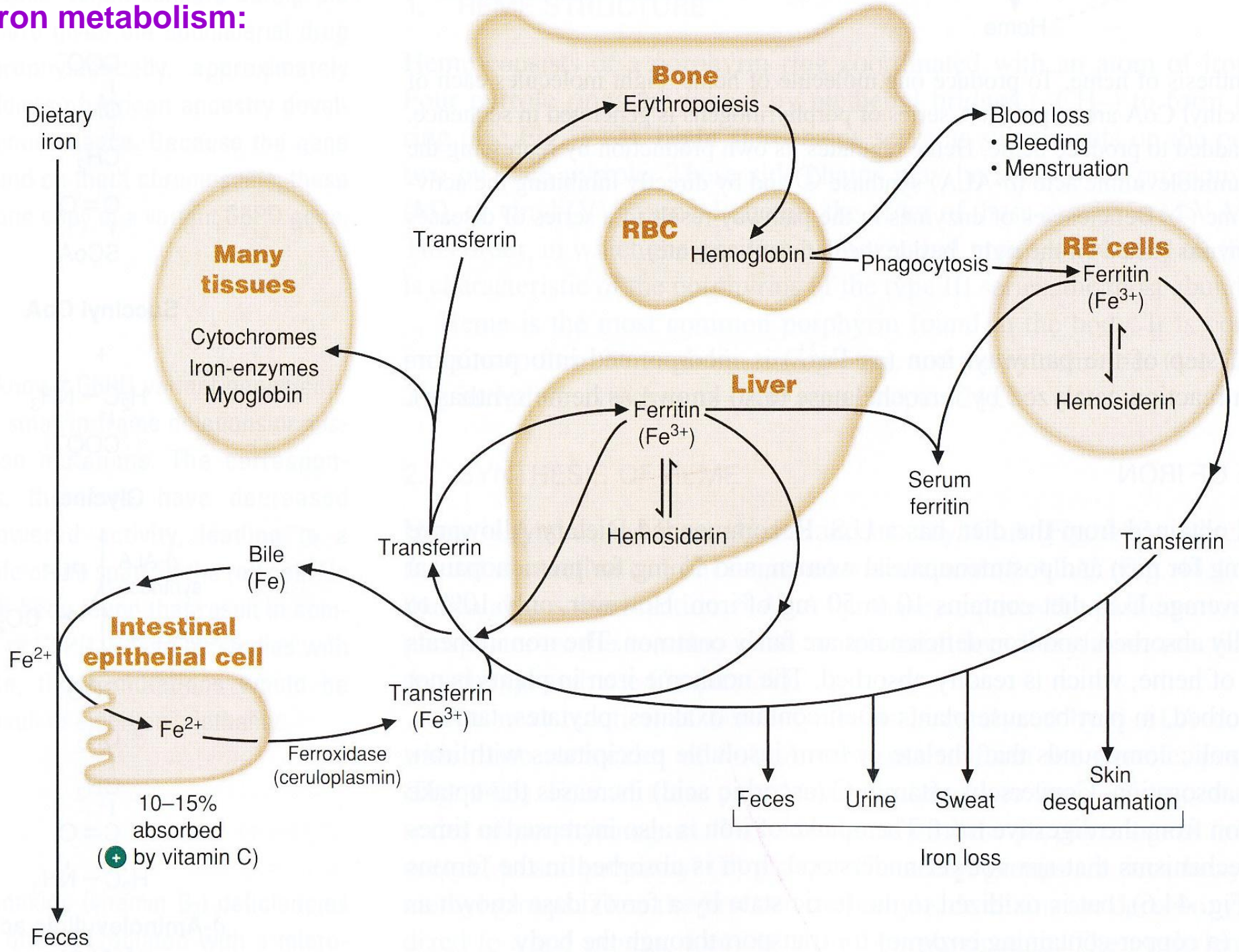


inhibited by lead



Porphobilinogen
(a pyrrole)

Iron metabolism:



Iron metabolism:

- Recommended dietary allowance 10-15 mg (only 10-15% is normally absorbed)

Iron distribution:

	g	%
Hemoglobin	2,5	68
Myoglobin	0,15	4
Transferrin	0,003	0,1
Ferritin, tissue	1,0	27
Ferritin, serum	0,0001	0,004
Enzymes	0,02	0,6
Total	3,7	100

Intestinal absorption of iron:

- in the duodenum
 - regulation (by the synthesis of apoferritin within mucosal cells)
1. The heme iron (unknown mechanism)
 2. The nonheme iron
 - is not readily absorbed (chelates with oxalates, phytates, etc.)
 - vit. C increases the uptake

Iron transport:

- **Transferrin** (Fe^{3+}) $\text{Transferrin} + \text{Fe}^{3+} + \text{CO}_3^{2-} \rightarrow \text{Transferrin} \cdot 2(\text{Fe}^{3+}\text{CO}_3^{2-})$
 - only one third saturated with iron
 - unsaturated transferrin protects against infections (iron overload and infection)
- **Lactoferrin**
 - binds iron in milk
 - antimicrobial effect (protects newborns from gastrointestinal infections)
- **Haptoglobin**
 - binds hemoglobin in the plasma

Iron storage:

- **Ferritin** (Fe^{3+})
 - storage of iron (hepatocytes, RES, muscles)
 - in the blood → sensitive indicator of the amount of iron in the body
- **Hemosiderin**
 - when iron is in excess (amorphous iron deposition)

Iron-containing proteins:

1. Heme proteins

Hemoglobin

Myoglobin

Enzymes that contain heme as their prosthetic group (catalase, peroxidase, NO synthase)

2. Nonheme proteins

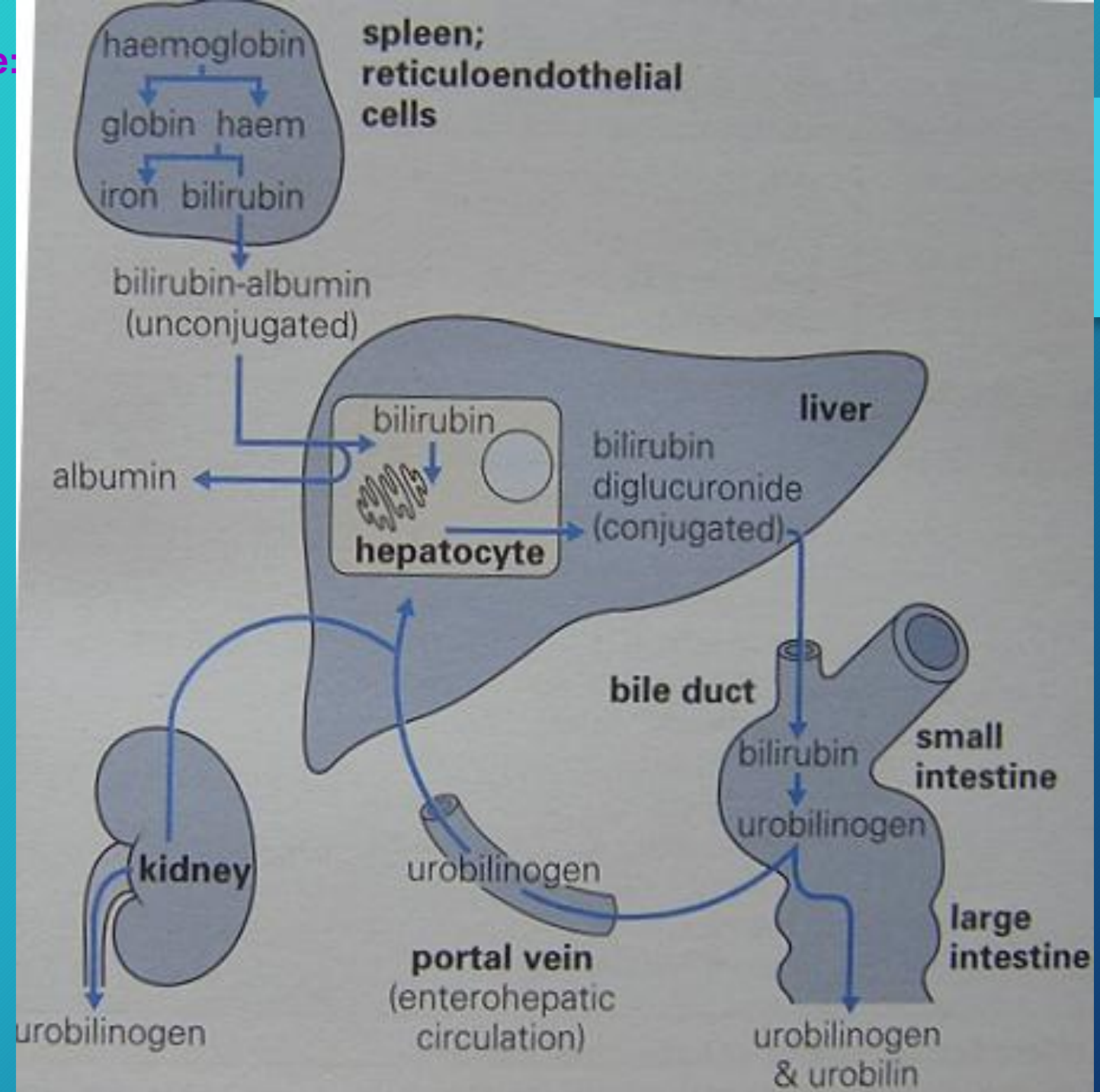
Transferrin

Ferritin

Enzymes that contain iron at the active site

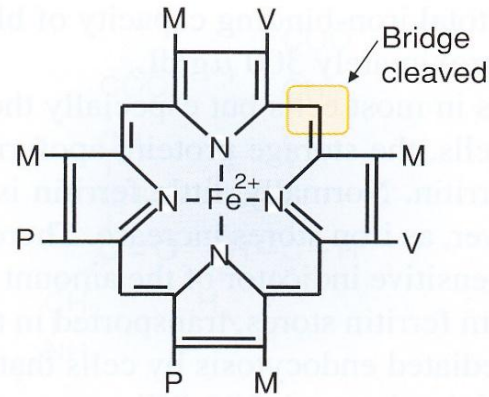
Iron-sulphur proteins

Degradation of heme:

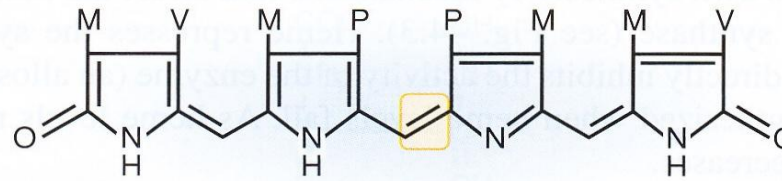
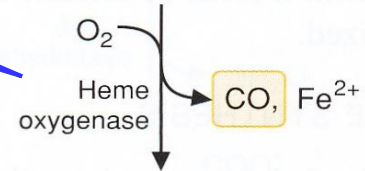


Conversion of heme to bilirubin:

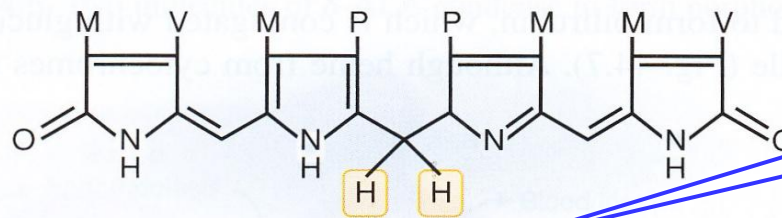
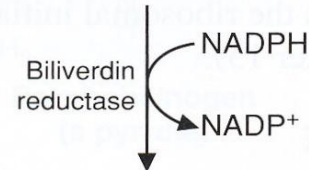
ER enzyme system



Heme



Biliverdin IX α



Bilirubin IX α

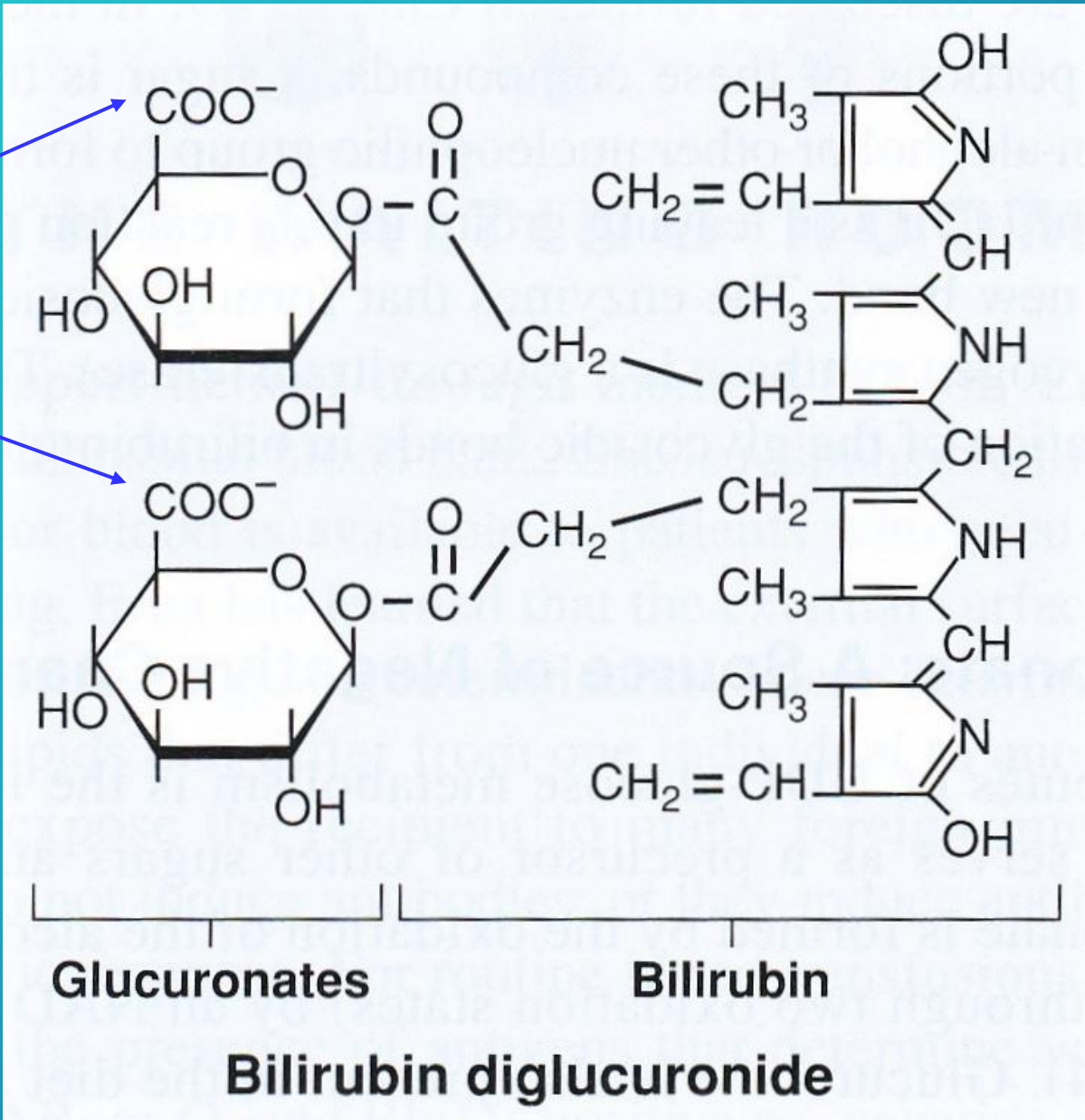
the major source is Hg

Cytoprotective role:

- CO
- biliverdin

Formation of bilirubin diglucuronide:

increase the water solubility of bilirubin

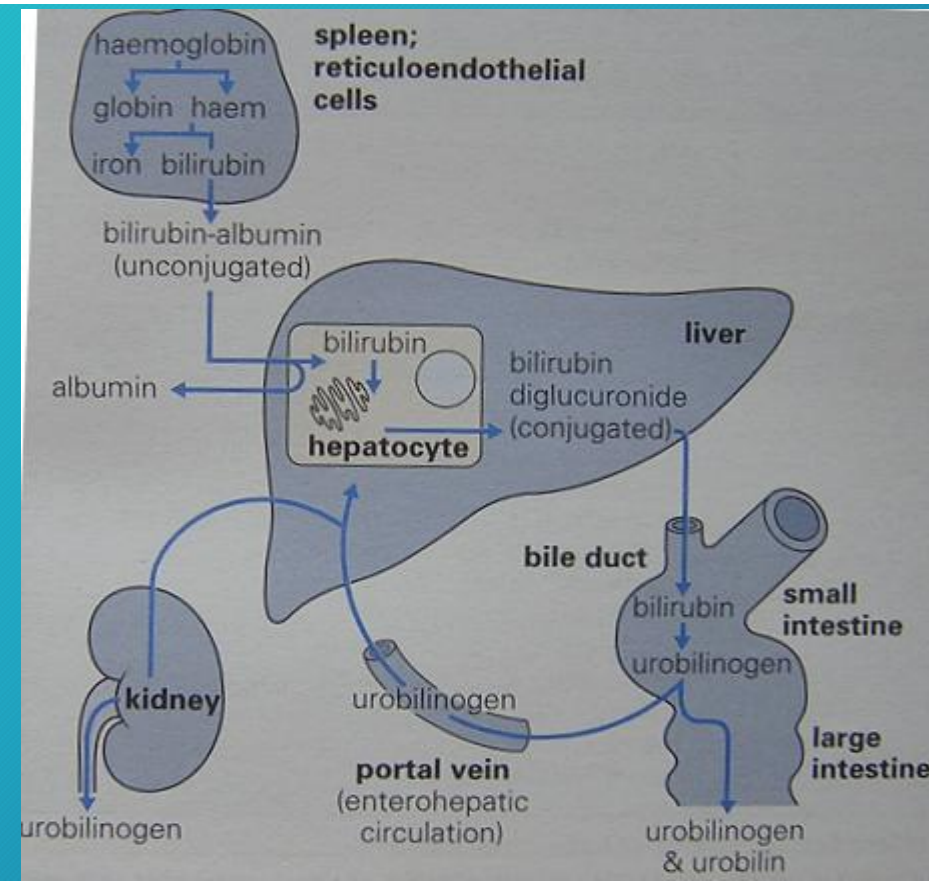


Hyperbilirubinemia

- Elevated bilirubin levels in the blood (>10 mg/l); bilirubin may diffuse into peripheral tissues, giving them a yellow color (jaundice)
- Cause:
 - 1. Pre-hepatic:** excessive formation of bilirubin by increased degradation of erythrocytes (icterus neonatus, hemolytic anemia)
 - 2. Hepatic:** insufficient processing of bilirubin as a result of liver defects (hepatitis, liver toxic damage, cirrhosis, hepatic failure)
 - 3. Post-hepatic:** by impaired excretion of gall (obstructive jaundice due to gallstones, inflammation of biliary tract)
- Unconjugated bilirubin can cross the blood-brain barrier, leading to brain damage
- Jaundice in neonates (increased bilirubin degradation+immaturity of the conjugation enzymes): phototherapy – isomerization of bilirubin to more soluble pigments

Type	Cause	Example	Frequency
Prehepatic	Hemolysis	Autoimmune Haemoglobinopathy	Rare According to the region
Hepatic	Infection Damage Genetics Autoimmune Newborn	Hepatitis A,B,C Alcohol, drugs Gilbert´s syndrome Wilson´s disease α_1 -Antitrypsin deficiency Chronic hepatitis Physiologic	Very common Common 1 in 20 1 in 200 000 1 in 1000 Rare Very common
Posthepatic	Intrahepatic bile ducts Extrahepatic bile ducts	Drugs Primary biliary cirrhosis Cholangitis Gallstones Pancreatic cancer	Common Rare Common Very common Rare

	Bilirubin			Urobilinogen	
	blood	urine	deriv. in feces	blood	urine
Prehepatic	↑↑(UC)	N	↑	↑	↑
Intrahepatic	↑↑(both)	↑	N	↑↑	↑↑
Posthepatic	↑↑(C)	↑	↓	↓	↓



METABOLISM OF BILE PIGMENTS IN HEPATIC JAUNDICE

